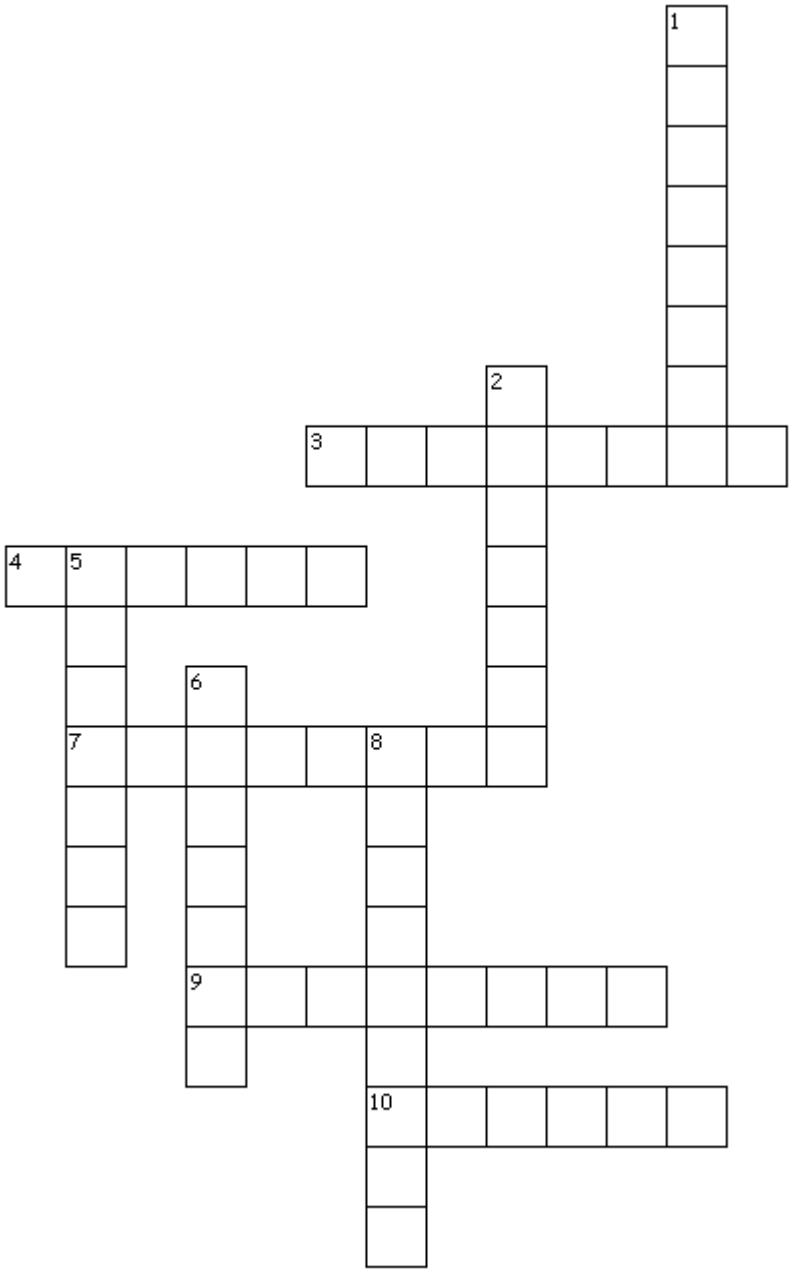
Draw the Bohr structure of all the following

Hydrogen (H)	Helium (He)	Lithium (Li)	Beryllium (Be)	Boron (B)
Carbon (C)	Nitrogen (N)	Oxygen (O)	Fluorine (F)	Neon (Ne)
Sodium (Na)	Magnesium (Mg)	Aluminium (Al)	Silicon (Si)	Phosphorus (P)
Sulfur (S)	Chlorine (Cl)	Argon (Ar)	Potassium (K)	Calcium (Ca)

Q	I	N	V	P	I	P	F	I	R	E	M	C	Z	K
Z	S	H	J	L	R	N	U	O	J	E	M	I	B	S
B	S	N	A	P	Q	O	W	O	N	C	N	D	T	X
J	U	K	E	E	A	A	P	D	R	C	I	O	I	N
S	L	O	A	G	T	D	E	E	S	G	L	I	E	S
A	L	R	B	R	O	L	O	V	R	Q	C	R	X	X
H	T	C	M	S	E	L	M	A	R	T	N	E	P	P
H	J	W	S	E	W	E	A	E	E	Z	I	P	L	E
T	U	W	V	N	L	O	I	H	T	T	R	E	O	R
C	O	L	U	M	N	E	S	E	S	A	G	W	S	I
E	I	E	I	M	P	I	A	S	J	O	L	H	I	O
L	L	P	O	B	S	R	B	D	P	U	N	S	V	D
B	P	T	B	G	O	T	G	I	A	Y	J	D	E	N
O	A	E	S	T	N	E	M	E	L	E	V	N	Z	C
N	S	J	N	K	G	W	P	W	N	I	D	U	I	J

ALKALI
EARTH
FIRE
HALOGENS
METALS
PERIODIC
SONG
ATOM
ELEMENTS
GASES
LEAD
NOBLE
PROPERTIES
COLUMN
EXPLOSIVE
GROUP
MENDELEEV
PERIOD
ROW

The periodic table and atomic structure



Across
3. element with the atomic number 1?
4. these group one metals are called?
7. name for group 7 non-metals.
9. these subatomic partcles have no charge.
10. group one metals react very violently with water and o-----.
Down
1. group two metals are called -------- earth metals?
2. positive parts of an atom?
5. element with chemical symbol Li?
6. substances that contain only one type of atom. e-----t.
8. these have a negative charge.